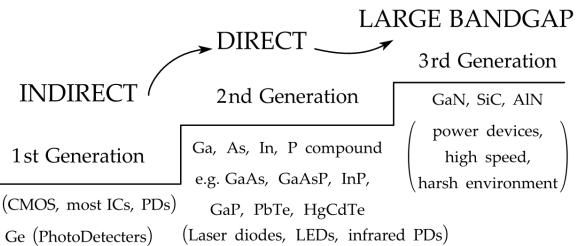
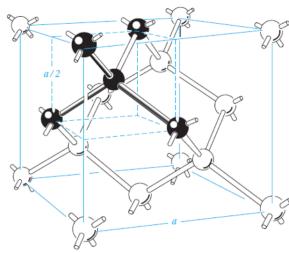


# Semiconductor Midterm CheatSheet (chapter1–4)

Boltzmann's constant ppm/ppb/ppt	$e = 1.602 \times 10^{-19} C$ $k = 8.62 \times 10^{-5} eV/K = 1.38 \times 10^{-23} J/K$ partpermillion/billion/trillion
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Si	Ge
$E_g(Si) = 1.1\text{eV}$	$E_g(Ge) = .67\text{eV}$
$5 \times 10^{22}\text{atoms/cm}^3$	
$n_i \approx 1.5 \times 10^{10}\text{cm}^{-3}$	$n_i = 2.4 \times 10^{13}\text{cm}^{-3}$
$\mu_n = 1350\text{cm}^3/\text{V-s}, \mu_p = 480\text{cm}^3/\text{V-s}$	$\mu_n = 3900\text{cm}^3/\text{V-s}, \mu_n = 1900\text{cm}^3/\text{V-s}$

three types of solids: Crystalline(silicon wafer), Polycrystalline(Poly-Si), Amorphous(glass)  
 Primitive Cell, Unit Cell, Miller Indices ( $hkl$ ) Diamond Lattice, Zinc Blende Structure,  
 Wurtzite Structure. packing fraction =  $\frac{V_{\text{atom}} \cdot N_{\text{atom}}}{V_{\text{cell}}} = \frac{\frac{4}{3}\pi(\frac{d_{\text{atom,min}}}{2})^3 \cdot N_{\text{atom}}}{(\text{lattice constant})^3}$



Czochralski Growth (CZ) for Si:

$$k_d = \frac{C_s}{C_l} = \frac{[\text{impurity in solid}]}{[\text{impurity in liquid}]}, \text{MASS} = \frac{[\text{number of atoms}]}{(\text{Avogadro's constant})} (\text{atomic mass})$$

Liquid Encapsulated Czochralski Growth (LEC), Bridgman Growth (BR) for GaAs

## Atom structure

Heisenberg Uncertainty Principle:  $(\Delta x)(\Delta p_x) \geq \hbar/2$ ,  $(\Delta E)(\Delta t) \geq \hbar/2$  The probability sum of finding a particle over the entire space  $\int_{-\infty}^{\infty} \Psi^* \Psi dx dy dz = 1$ . Average value  $\langle Q \rangle = \int_{-\infty}^{\infty} \Psi^* Q_{\text{op}} \Psi dx dy dz$ , where  $Q_{\text{op}}$  is defined as,

classical variable	quantum operator
$x, f(x)$	$x, f(x)$
momentum $p(x)$	$\frac{\hbar}{j} \frac{\partial}{\partial x}$
energy $E$	$-\frac{\hbar}{j} \frac{\partial}{\partial t}$

Schrödinger equation  $E_k(\frac{p^2}{2m}) + E_p(V) = E_{\text{total}}(E)$ , by replacing  $p$  and  $E$  with operators, we can derive  $-\frac{\hbar^2}{2m} \frac{\partial^2 \psi(x)}{\partial x^2} \phi(t) + V(x)\psi(x)\phi(t) = -\frac{\hbar}{j} \psi(x) \frac{\partial \phi(t)}{\partial t}$ .

Quantum Numbers for $\Psi(r, \theta, \phi) = R_n(r)\Theta_l(\theta)\Phi_m(\phi)$		
$n$	principal quantum number, shell	1, 2, ...
$l$	angular momentum quantum number, subshell	$0, 1, \dots, n-1$ ( $s, p, d, f, g, \dots$ )
$m$	magnetic quantum number, orientation of orbitals (within the subshells)	$0, \pm 1, \dots, \pm(n-1)$
$s$	electron spin	$\pm \frac{1}{2}$

**Pauli exclusion principle** No two electrons in an interacting system with the same  $n, l, m, s$ . The only two electrons with same  $n, l, m$  must have opposite spins

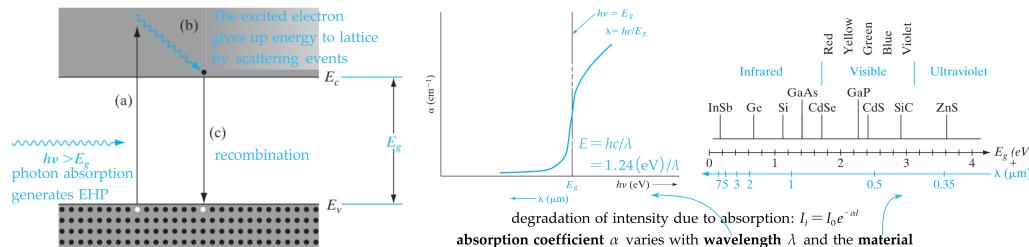
Atomic number (Z) Element	1s	n = 1		2		3		4		Shorthand notation
		l = 0	0 1	0 1 2	0 1	4s 4p				
1 H	1	1				1s <sup>1</sup>				
2 He	2					1s <sup>2</sup>				
3 Li		1				1s <sup>2</sup> 2s <sup>1</sup>				
4 Be		2				1s <sup>2</sup> 2s <sup>2</sup>				
5 B		2	1			1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>1</sup>				
6 C		2	2			1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>2</sup>				
7 N		2	3			1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup>				
8 O		2	4			1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>4</sup>				
9 F		2	5			1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>5</sup>				
10 Ne		2	6			1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup>				
11 Na			1			[Ne] 3s <sup>1</sup>				
12 Mg			2			3s <sup>2</sup>				
13 Al			2	1		3s <sup>2</sup> 3p <sup>1</sup>				
14 Si			2	2		3s <sup>2</sup> 3p <sup>2</sup>				
15 P			2	3		3s <sup>2</sup> 3p <sup>3</sup>				
16 S			2	4		3s <sup>2</sup> 3p <sup>4</sup>				
17 Cl			2	5		3s <sup>2</sup> 3p <sup>5</sup>				
18 Ar			2	6		3s <sup>2</sup> 3p <sup>6</sup>				
19 K				1		[Ar] 4s <sup>1</sup>				
20 Ca				2		4s <sup>2</sup>				
21 Sc				2	2	3d <sup>1</sup> 4s <sup>2</sup>				
22 Ti				3	2	3d <sup>2</sup> 4s <sup>2</sup>				
23 V				5	1	3d <sup>3</sup> 4s <sup>2</sup>				
24 Cr				5	2	3d <sup>5</sup> 4s <sup>2</sup>				
25 Mn				6	2	3d <sup>6</sup> 4s <sup>2</sup>				
26 Fe				7	2	3d <sup>7</sup> 4s <sup>2</sup>				
27 Co				8	2	3d <sup>10</sup> 4s <sup>2</sup>				
28 Ni				10	1	3d <sup>10</sup> 4s <sup>2</sup>				
29 Cu				10	2	3d <sup>10</sup> 4s <sup>2</sup>				
30 Zn				10	2	3d <sup>10</sup> 4s <sup>2</sup>				
31 Ga				10	2	3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>1</sup>				
32 Ge				10	2	3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>2</sup>				
33 As				10	2	3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>3</sup>				
34 Se				10	2	3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>4</sup>				
35 Br				10	2	3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>5</sup>				
36 Kr				10	2	3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>6</sup>				

n	l	m	s/h	Allowable states in subshell	Allowable states in complete shell
1	0	0	$\pm \frac{1}{2}$	2	2
2	0	0	$\pm \frac{1}{2}$	2	8
	-1	0	$\pm \frac{1}{2}$	6	
	1	0	$\pm \frac{1}{2}$	6	
3	0	0	$\pm \frac{1}{2}$	2	18
	-1	0	$\pm \frac{1}{2}$	6	
	1	0	$\pm \frac{1}{2}$	6	
2	-2	0	$\pm \frac{1}{2}$	10	18
	-1	0	$\pm \frac{1}{2}$	10	
	1	0	$\pm \frac{1}{2}$	10	
2	-2	1	$\pm \frac{1}{2}$	10	18
	-1	1	$\pm \frac{1}{2}$	10	
	1	1	$\pm \frac{1}{2}$	10	
2	-2	2	$\pm \frac{1}{2}$	10	
	-1	2	$\pm \frac{1}{2}$	10	
	1	2	$\pm \frac{1}{2}$	10	
2	-2	3	$\pm \frac{1}{2}$	10	
	-1	3	$\pm \frac{1}{2}$	10	
	1	3	$\pm \frac{1}{2}$	10	
2	-2	4	$\pm \frac{1}{2}$	10	
	-1	4	$\pm \frac{1}{2}$	10	
	1	4	$\pm \frac{1}{2}$	10	
2	-2	5	$\pm \frac{1}{2}$	10	
	-1	5	$\pm \frac{1}{2}$	10	
	1	5	$\pm \frac{1}{2}$	10	
2	-2	6	$\pm \frac{1}{2}$	10	
	-1	6	$\pm \frac{1}{2}$	10	
	1	6	$\pm \frac{1}{2}$	10	

Silicon:  $1s^2 2s^2 2p^6 3s^2 3p^2$

carrier concentrations = (Fermi distribution)(available states density)	$n_0 = \int_{E_c}^{\infty} f(E) N(E) dE = N_c f(E_c) \approx N_c e^{-(E_c - E_F)/kT}$
$p_0 = \int_0^{E_v} (1 - f(E)) N(E) dE = N_v (1 - f(E_v)) \approx N_v e^{-(E_F - E_v)/kT}$	
intrinsic $\begin{cases} n_i = \sqrt{N_c N_v} e^{-E_g/2kT} \\ n_0 p_0 = n_i^2 \end{cases}$	doped, equilibrium $\begin{cases} n_0 = n_i e^{(E_F - E_i)/kT} \\ p_0 = n_i e^{(E_i - E_F)/kT} \end{cases}$
excess carriers, quasi-Fermi level $\begin{cases} n = n_i e^{(F_n - E_i)/kT} \\ p = n_i e^{(E_i - F_p)/kT} \end{cases}$	(for majority carriers $F_n/F_p \approx E_i$ )

# Excess Carriers



Fluorescence (direct recombination) most energy directly into photon, fast( $10^{-10}\text{s}$ )

Phosphorescence (indirect recombination) needs extra (deep) energy level between  $E_g$ , gives smaller  $h\nu$ , slow(s); application: UV excites phosphors to generate white light instead of smooth spectrum

equilibrium no stimulate

steady state can have stimulation inputs, stabilises after a long time

## Recombination and Generation Balance

excess carrier concentrations  $\delta n(t) = \delta p(t)$ , initial values  $\Delta n = \Delta p$ .

low-level injection (linear)	n-type	p-type
$\frac{d\delta n(t)}{dt} = -\alpha_r(n_0 + p_0)\delta n(t)$	$\delta n(t) = \Delta n e^{-t/\tau}$	
minority carrier lifetime $\tau_n = \frac{1}{\alpha_r(n_0 + p_0)}$	$\tau_n = (\alpha_r n_0)^{-1}$	$\tau_p = (\alpha_r p_0)^{-1}$
$g_{op} = \alpha_r[(n_0 + p_0)\delta n(t) + \delta n^2(t)]$	$\delta n = \delta p = g_{op}\tau_n$	

photoconductive decay:

$$\sigma(t) = q[n(t)\mu_n + p(t)\mu_p]$$

$$\Delta\sigma = qg_{op}(r_n\mu_n + r_p\mu_p) = q(\delta n\mu_n + \delta p\mu_p)$$

Direct Recombination:

net change = thermal generation – recombination

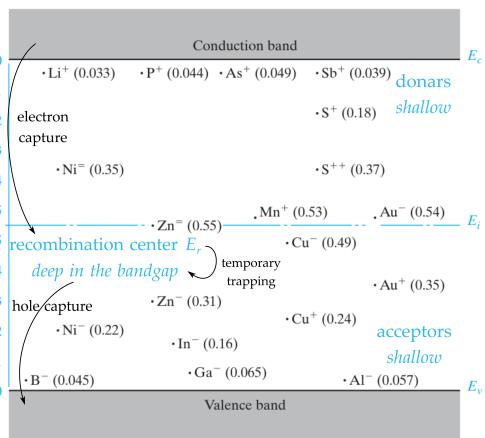
$$\begin{aligned} \frac{dn(t)}{dt} &= \alpha_r n_i^2 - \alpha_r n(t)p(t) \\ &= -\alpha_r[(n_0 + p_0)\delta n(t) + \delta n^2(t)] \end{aligned}$$

steady state EHP G-R balance:

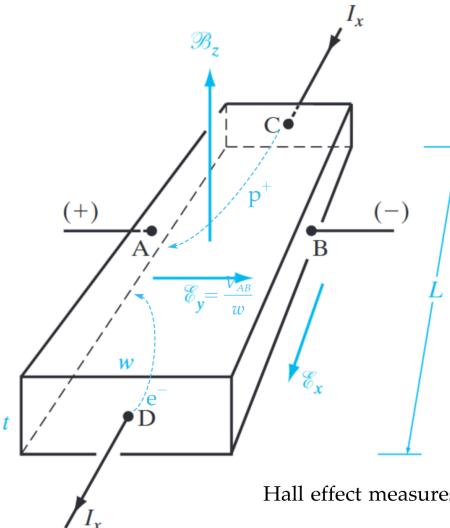
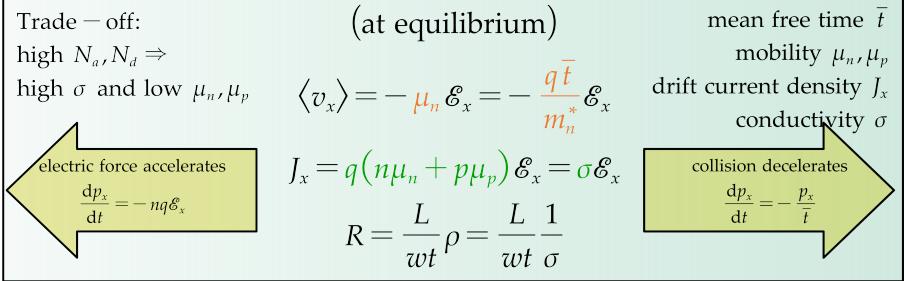
generation = recombination

$$[\text{equilibrium}]G(T) = \alpha_r n_0 p_0 = \alpha_r n_i^2$$

$$[\text{steady state}]G(T) + g_{op} = \alpha_r n_0 p_0 + \alpha_r[(n_0 + p_0)\delta n(t) + \delta n^2(t)]$$



## Drift and Diffusion



Hall effect measures majority carrier mobility

